

A PHOTOCHEMICAL APORPHINE SYNTHESIS

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Aporphine alkaloids have been synthesized previously only by the Pschorr cyclization of 1-(*o*-aminobenzyl)tetrahydroisoquinolines.¹ We now wish to report a fundamentally new aporphine synthesis in which a structurally simpler aporphine precursor is used, and in which the key synthetic step is a special application of the stilbene-phenanthrene photocyclization.² The method is illustrated here by a new synthesis of dl-nuciferine (V).

The reaction of 1-benzyl-6,7-dimethoxy-3,4-dihydroisoquinoline (I)³ with ethyl chloroformate gave, in 67% yield (calculated on unrecovered starting material), 1-benzylidene-2-carbomethoxy-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline (II), m.p. 132-134°, λ_{\max} 227 m μ ($\epsilon = 23,000$) and 328 (25,000).⁴ Compound II was assigned the *trans* configuration of the aromatic substituents about the double bond, since its NMR spectrum revealed considerable shielding of the methyl of the carbomethoxy group (triplet at δ 0.78) by the benzylidene phenyl.

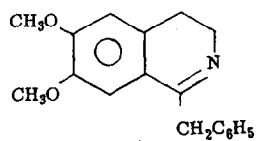
Ultraviolet irradiation of II in ethanol solution in the presence of iodine (0.6 equivalents) gave, in 15% yield, N-carbomethoxy-6a,7-dehydronornuciferine (III), m.p. 128-130°, λ_{\max} 256 m μ ($\epsilon = 39,000$), 263 (44,000), 311 (11,500), 323 (11,500), 356 (1,800), and 375 (1,900). The NMR spectrum of III, which was fully consistent with the assigned structure, showed as points of special interest a normal carbomethoxy methyl (triplet at δ 1.31) and the expected low field proton at C-11 (δ 9.58) of the aporphine system.

Lithium aluminum hydride reduction of III in ether solution afforded, in 70% yield, 6a,7-c-hydronuciferine (IV), m.p. 126-130°, λ_{max} 253 m μ (ϵ 93,000), 264 (96,000), 293 (18,000), and 327 (31,000). The NMR of IV showed, as features of major significance, the low field C-11 proton at δ 9.48, and a 3-proton singlet at δ 3.02 attributable to the N-methyl group. Compound IV was identical (IR, UV, m.p. and R_f) with material obtained by the oxidation of synthetic nuciferine⁵ with potassium permanganate in acetone.⁶

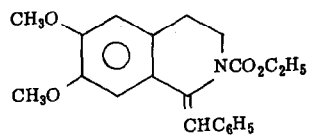
Catalytic reduction (PtO₂ catalyst) of IV in acetic acid solution afforded, in 70% yield, dl-nuciferine (V), identical in all respects (IR, UV, m.p. and R_f) with authentic nuciferine.⁵

Work is in progress to extend this new aporphine synthesis to more complex examples, and to improve the overall yield of the process.⁷

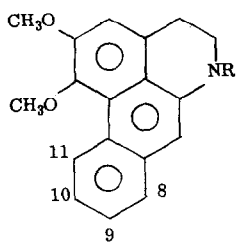
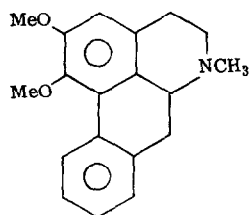
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I



II

III, R = $\text{CO}_2\text{C}_2\text{H}_5$ IV, R = CH_3 

V

REFERENCES

1. For a brief review of the synthesis of aporphines, see A. R. Pinder, in E. H. Rodl, Chemistry of Carbon Compounds, Elsevier Publishing Co., N. Y. (1960), Vol. IV, Chap. 25.
2. The following references describe some recent examples of this reaction:
(a) G. J. Fonken, Chem. and Ind. (London), 36, 1327 (1962); (b) F. B. Mallory, C. S. Wood, and J. T. Gordon, J. Am. Chem. Soc., 86, 3094 (1964); (c) W. Carruthers and H. N. M. Stewart, J. Chem. Soc., 6221 (1965); (d) D. J. Collins and J. J. Hobbs, Chem. and Ind. (London), 39, 1725 (1965); and (e) K. A. Muszkat, D. Gegiou, and E. Fischer, Chem. Comm., 447 (1965).
3. G. Tsatsas, Ann. pharm. franc., 10, 61 (1952); Chem. Abstr., 46, 11208 (1952).
4. Melting points are uncorrected. Satisfactory analyses were obtained for all new compounds reported. Irradiations were carried out using a model 679A36 Hanovia lamp and a quartz probe. Ultraviolet spectra were determined in cyclohexane; NMR spectra were determined in CDCl₃.
5. J. A. Weisbach and B. Douglas, J. Org. Chem., 27, 3738 (1962).
6. The first report of the oxidation of an aporphine (thalicarpine) to a dehydroaporphine by this method appeared recently: N. M. Mollov and H. B. Dutschewska, Tetrahedron Letters, 853 (1966).
7. For an independent report of the results similar to ours, see the accompanying communication of N. C. Yang, et al.